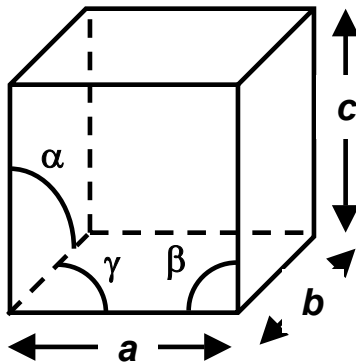


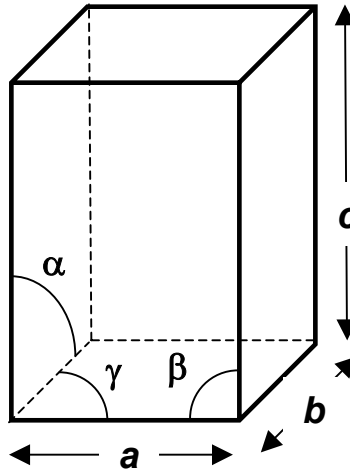
**Seven unique Bravais lattice classes based on point symmetry
(subset of 14 total unique Bravais lattices)**

Cubic



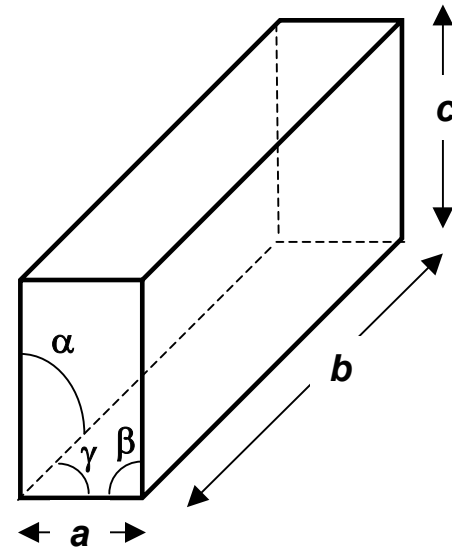
$$a=b=c$$
$$\alpha=\beta=\gamma=90^\circ$$

Tetragonal



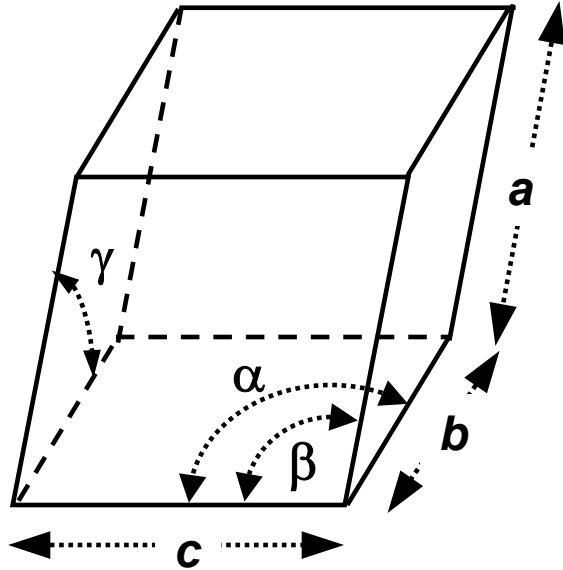
$$a=b \neq c$$
$$\alpha=\beta=\gamma=90^\circ$$

Orthorhombic



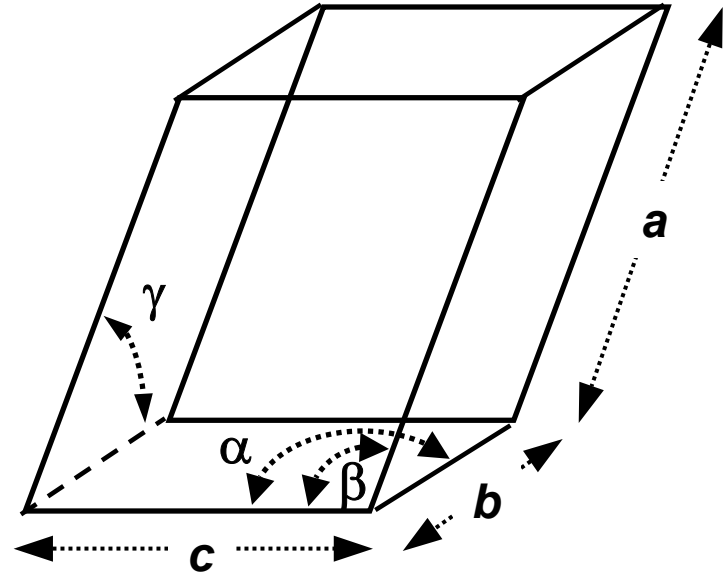
$$a \neq b \neq c$$
$$\alpha=\beta=\gamma=90^\circ$$

Monoclinic (six sided)



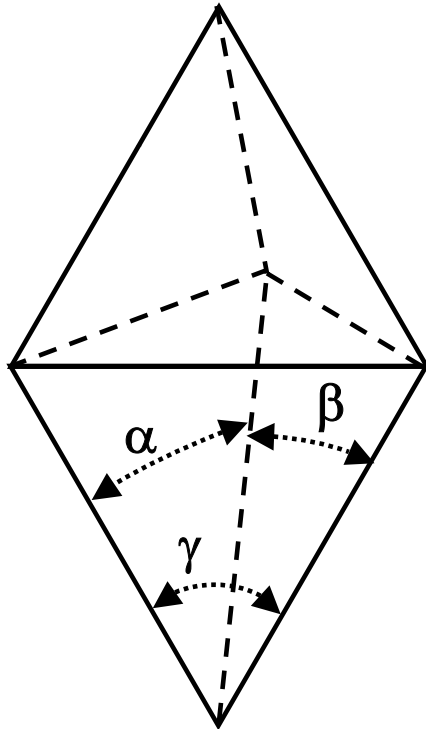
$$\begin{aligned} a &\neq b \neq c \\ \alpha &= \gamma = 90^\circ \\ \beta &\neq 90^\circ \end{aligned}$$

Triclinic (six sided)



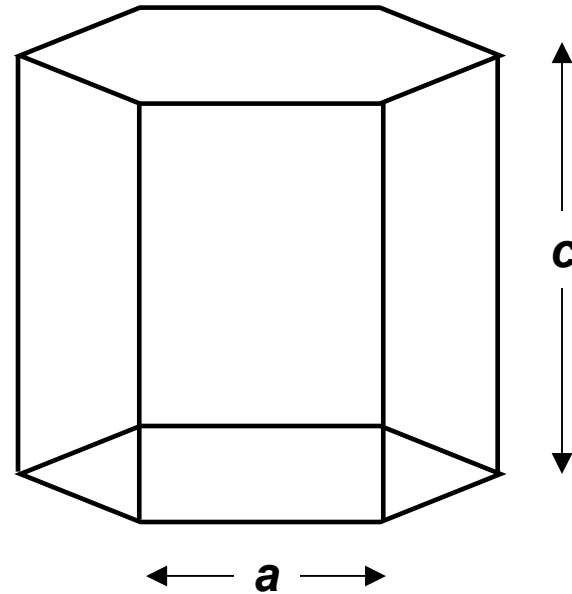
$$\begin{aligned} a &\neq b \neq c \\ \alpha &\neq \beta \neq \gamma \end{aligned}$$

Trigonal (six sided)

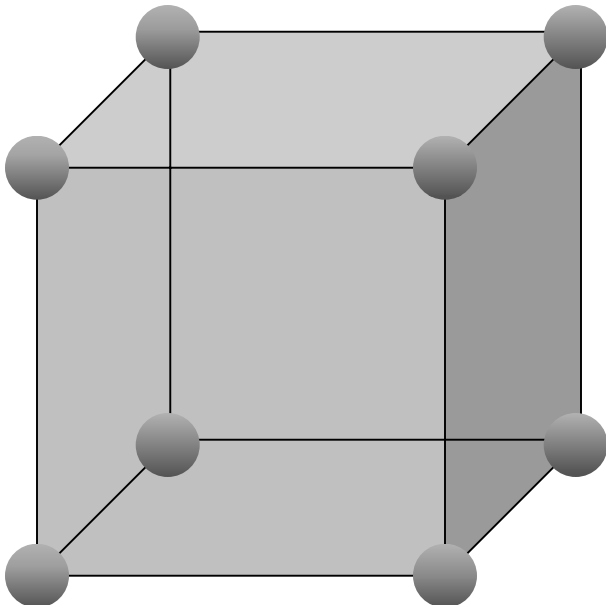


$$a = b = c$$
$$\alpha = \beta = \gamma < 90^\circ$$

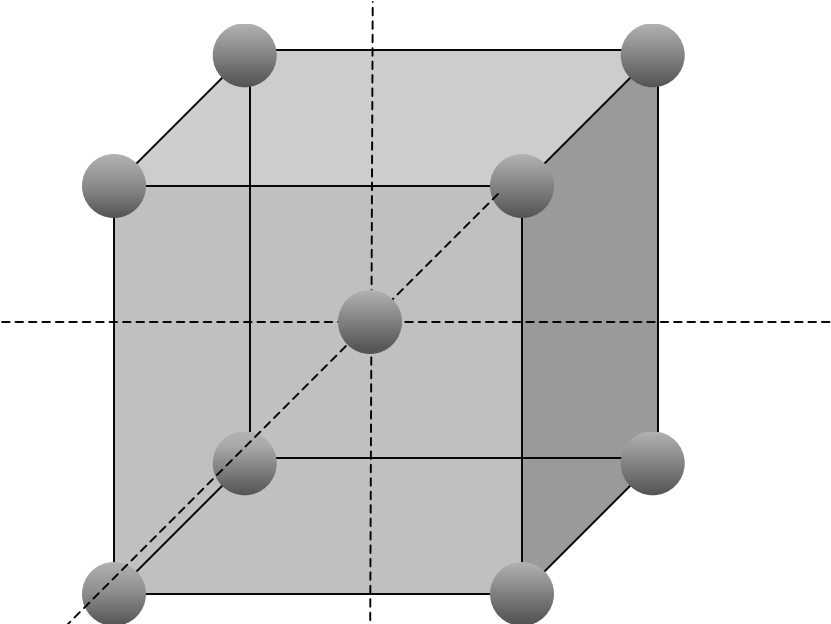
Hexagonal (8 sided)



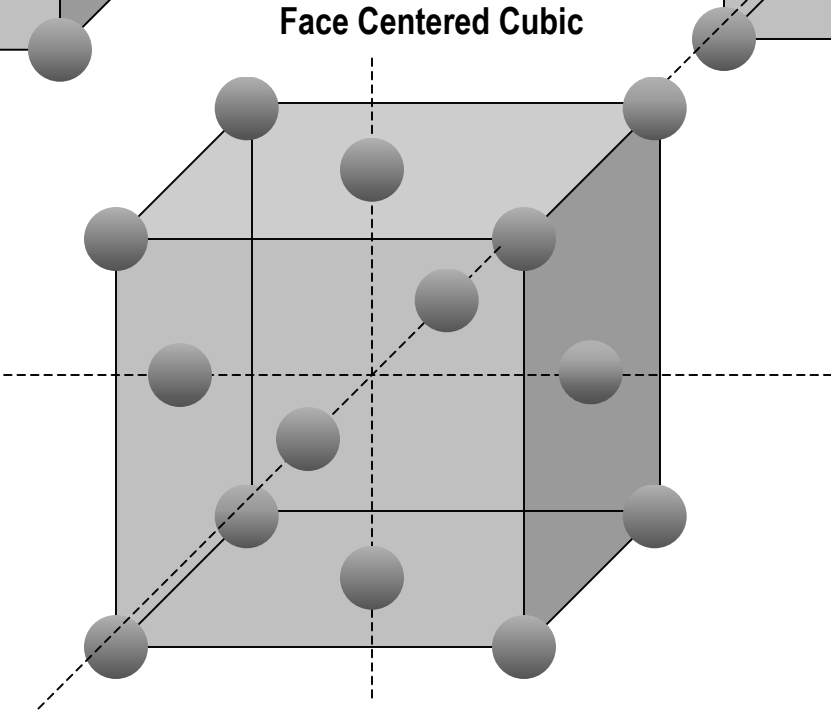
Cubic Crystal Class



Simple Cubic

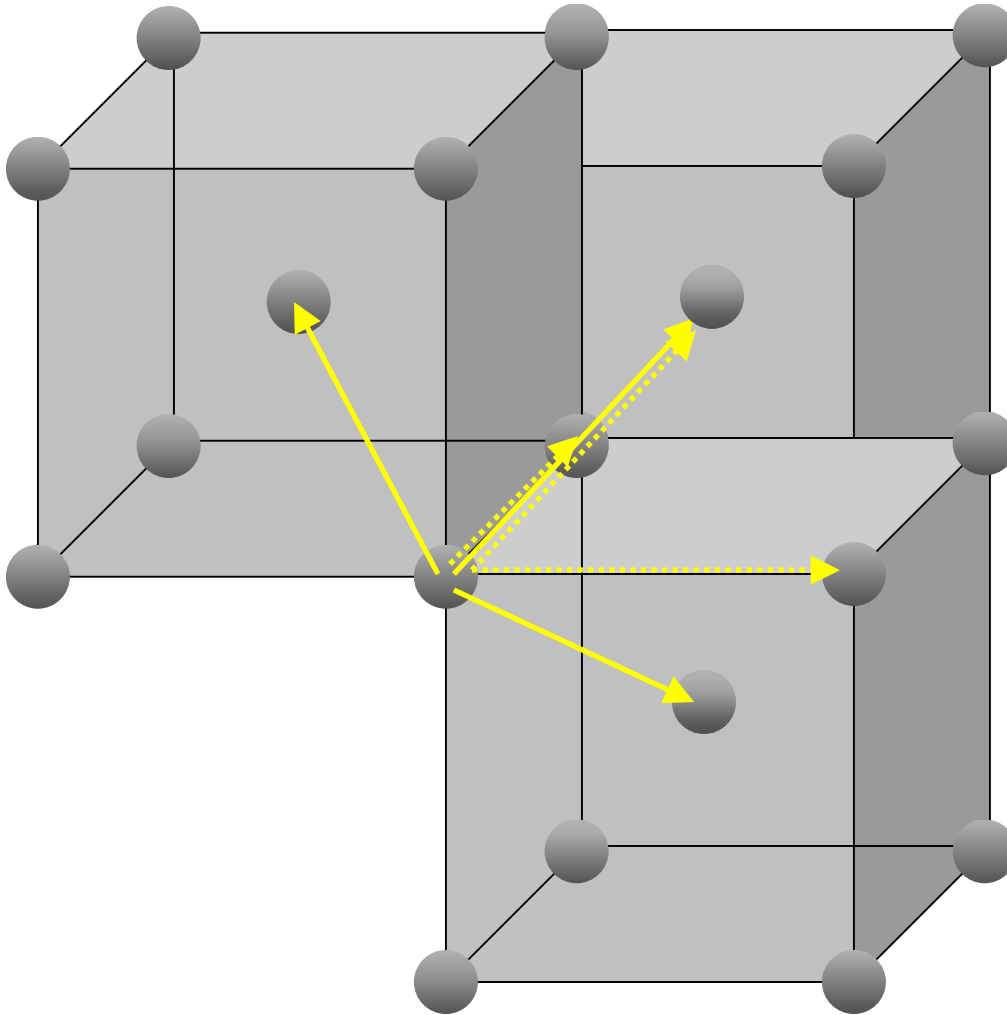


Body Centered Cubic



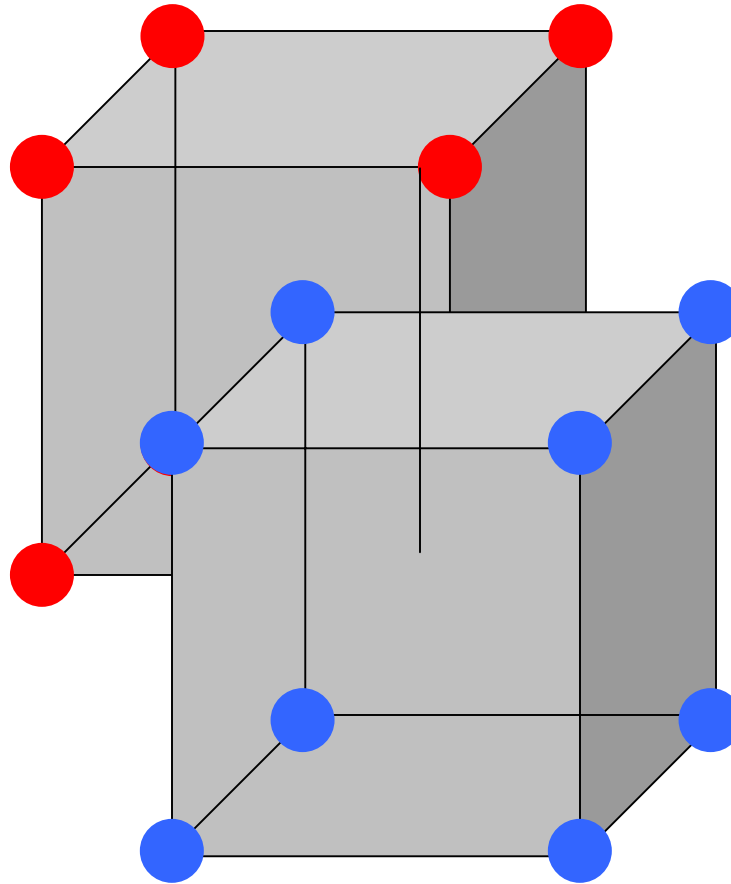
Face Centered Cubic

**Two Possible Sets of Primitive Lattice Vectors for bcc Lattice
(there are many others too)**



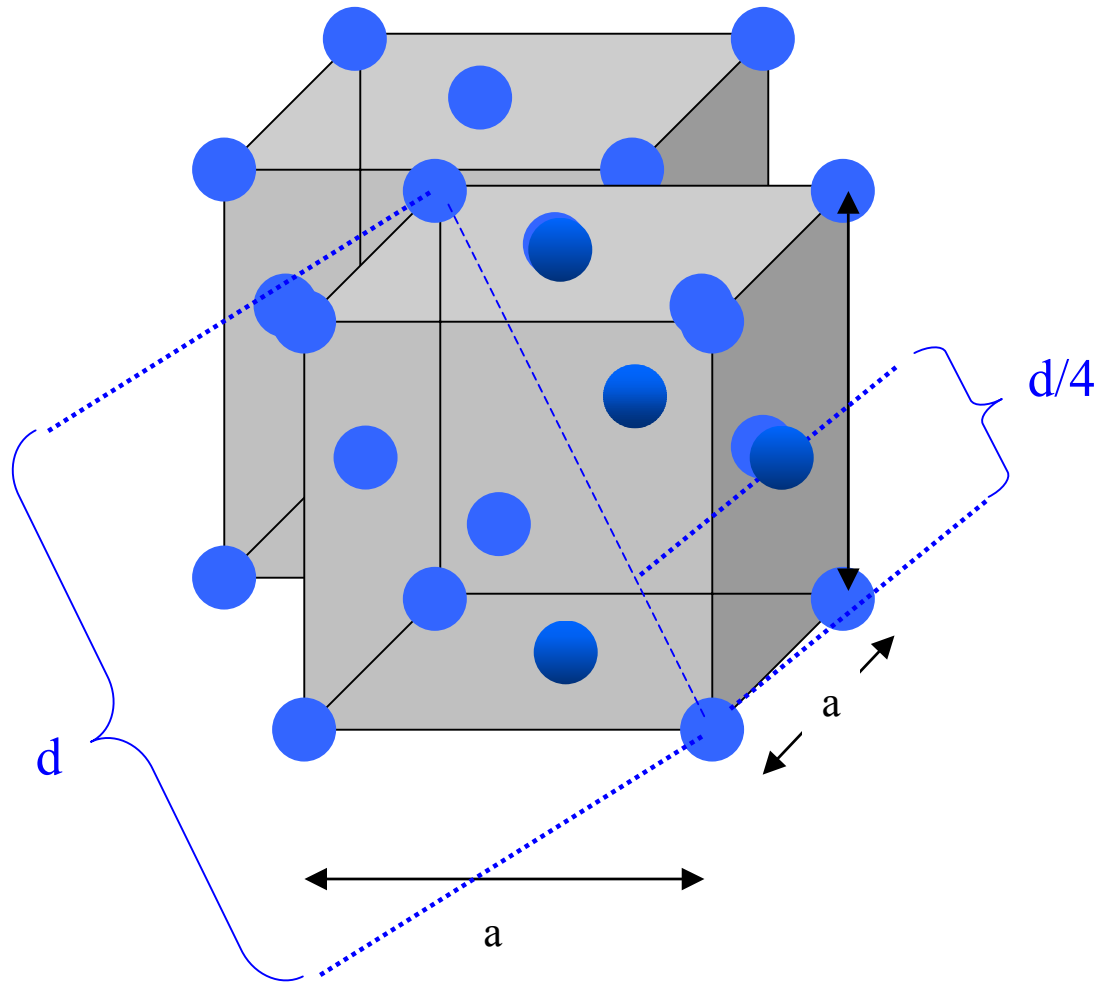
Lattice with a Basis (1)

CsCl Structure: Two interpenetrating bcc lattices, each with different atom



Lattice with a Basis (2)

Diamond Structure: Two interpenetrating fcc lattices, each with same atom



Lattice with a Basis (3)

Zincblende: Two interpenetrating fcc lattices with different atoms on each

